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46. A 10.0-milliliter sample of NaOH(aq) is neutralized by 40.0 milliliters of 0.50 M HCl. What is the molarity of the NaOH(aq)?

(1) 1.0 M (3) 0.25 M
(2) 2.0 M (4) 0.50 M

$M_1 V_1 = M_2 V_2$ $M = \frac{m}{L}$

50. Which balanced equation represents nuclear fusion?

(1) ${}^2_1\text{H} + {}^3_1\text{H} \rightarrow {}^4_2\text{He}$
(2) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
(3) ${}^6_3\text{Li} + {}^1_0\text{n} \rightarrow {}^3_1\text{H} + {}^4_2\text{He}$
(4) $\text{CaO} + \text{CO}_2 \rightarrow \text{CaCO}_3$

(1) nuclei of H-1 and H-2, only
(2) nuclei of H-1 and H-3, only
(3) nuclei of H-2 and H-3, only
(4) nuclei of H-1, H-2, and H-3

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