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Unit 4: Solution Chemistry
Content Outline: Molarity (4.5)

1. Molarity

A. Molar mass

1. Refers to a measurement of molar mass that is relative to its molecular weight.
2. Joseph's Number: 6.02×10^{23}
3. Find the molecular weight of a molecule using the Periodic Table and then weigh out that many grams of the substance and that amount is equal to 1 mole. (molar mass = 342 so I need 342 grams)

B. Molarity

1. There are billions upon billions of molecules in a solution. (usually water)

IF have 2

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