

# Download File PDF Solubility Rules For Aqueous Solutions

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**Solubility Rules**

- **Solubility** of solute: amount dissolved in given quantity of solvent at given temperature
- Major ideas:
  - Many salts dissociate into ions in aqueous solution
  - If solid forms from combination of selected ions in solution
    - Solid must contain anion part and cation part
    - Net charge on solid must be zero
  - Simple solubility rules used to predict products of reactions in aqueous solutions

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