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#Jenny



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#Rio



Cool! I'am really happy

#Markus Jensen



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#Diego Butler



so many fake sites. this is the first one which worked! Many thanks

How the Form Soluble Compounds	Exceptions	How the Form Insoluble Compounds	Exceptions
Group 1 ions (Li ⁺ , Na ⁺ , etc.)		carbonate (CO ₃ ²⁻)	when combined with Group 1 ions or ammonium (NH ₄ ⁺)
acetate (CH ₃ COO ⁻)		chromate (CrO ₄ ²⁻)	when combined with Group 1 ions, Ca ²⁺ , Mg ²⁺ , or ammonium (NH ₄ ⁺)
nitrate (NO ₃ ⁻)		phosphate (PO ₄ ³⁻)	when combined with Group 1 ions or ammonium (NH ₄ ⁺)
nitrate (NO ₃ ⁻) or ClO ₄ ⁻		sulfide (S ²⁻)	when combined with Group 1 ions or ammonium (NH ₄ ⁺)
hydrogencarbonate (HCO ₃ ⁻)		hydroxide (OH ⁻)	when combined with Group 1 ions, Ca ²⁺ , Sr ²⁺ , or ammonium (NH ₄ ⁺)
chloride (Cl ⁻)			
sulfate (SO ₄ ²⁻)	when combined with Ag ⁺ , Pb ²⁺ , and Hg ₂ ²⁺		
iodide (I ⁻ , Br ⁻)	when combined with Ag ⁺ , Pb ²⁺ , and Hg ₂ ²⁺		
halides (I ⁻ , Br ⁻)	when combined with Ag ⁺ , Pb ²⁺ , and Hg ₂ ²⁺		

HOW TO USE TABLE F:
Find one of the ions in your formula in either Column 1 or Column 3. There are

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