

#Jenny



Finally I get this ebook, thanks for all these I can get now!

#Rio



Cool! I'am really happy

#Markus Jensen



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My friends are so mad that they do not know how I have all the high quality ebook which they do not!

#Diego Butler



so many fake sites. this is the first one which worked! Many thanks

26 Saturated and Unsaturated Solutions-S
May 6, 2014, 10:38 AM

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Saturated and Unsaturated Solutions
Is there a limit to the amount of solute that will dissolve in a solvent?

Why?
We use solutions every day. People who wear contact lenses use "lens solutions" to store their contacts and keep them wet. Athletes who consume sports drinks after exercising benefit from the electrolytes in those solutions. This activity will explore whether or not there is a limit to how much of one substance can dissolve in another.

Model 1 – Saturated and Unsaturated Solutions

Unsaturated Solutions

	Beaker A	Beaker B
Number of dissolved particles	5	10
Number of solid particles	0	0

1.0 g of solute added 2.0 g of solute added

Saturated Solutions

	Beaker C	Beaker D	Beaker E
Number of dissolved particles	16	17	18
Number of solid particles	0	0	0

3.0 g of solute added 7.0 g of solute added 9.0 g of solute added

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